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Chemical Bonding - Practice Test Questions & Chapter Exam ...

Chemical compounds tend to form so that each atom, by gaining, losing, or sharing electrons, has an octet of electrons in its highest occupied energy level. a pair of electrons that is not involved in bonding and that belongs exclusively to one atom. double and triple bonds are referred to as multiple bonds; CHAPTER 6 REVIEW Chemical Bonding SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

Chemistry Test- Chemical Bonding: Chapter 6 Flashcards ...

Assessment Chapter Test A

CHAPTER 6 - CHEMICAL BONDING Chemical Bond - a link between atoms that holds them together in a compound. Why Bonding Occurs - usually to get to a lower energy state. Most atoms drop in energy when they form bonds with other atoms, because bonding allows them to get to an octet.

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CHAPTER 6 TEST: CHEMICAL BONDING REVIEW SHEET

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Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative.

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Terms in this set (...) Metallic bonds. The chemical bonding that results from the attraction between metal atoms and the surrounding "sea of electrons". Properties of Metallic Bonds. - high electrical & thermal conductivity. - absorb a wide range of light frequencies & are therefore shiny. - malleable. - ductile.

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Use the diagram and the table below to classify the bond for each listed substance as ionic, non polar covalent, or polar covalent. Write your answers on the answer sheet. Percent ionic character 92% 50% 5% 0% Difference in 3.3 1.7 0.3 0.0. Electronega- Ionic Polar covalent Nonpolar Covalent. tivity 16.

CHAPTER 6 TEST: CHEMICAL BONDING REVIEW SHEET

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Chemical Bond 6.1 Chemical Bond 6.1 Chemical Bond 6.1 Chemical Bond 6.1 When the highest occupied energy level... Describe the type of electron configuration that makes an atom... An ____ is a model of an atom in which each dot represent... Definition: A model of an atom in which each dot represents a... Definition:...

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test sample chapter 6 workbook questions chemical bonds ...

A hydrogen bond is a dipole - dipole attraction between a partially positive hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction.

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6 Chemical Bonding - Effingham County School District

Chemical bond. a mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together. Chemical formula. a formula that indicates the relative numbers of atoms of each kind in a chemical compound by using atomic symbols and numerical subscripts.

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Assessment Chapter Test A

Chapter 6 Test The Structure of Matter EXTENDED RESPONSE. PART 1 CHOOSE ONE OF THE FOLLOWING. (6 points) Circle the question you are answering. 1. How does the type of chemical bonds present in a substance affect the substance's properties? Give at least two examples. 2. Compare and contrast ionic and covalent bonds. Give an example

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Holt McDougal Modern Chemistry Chapter 6: Chemical Bonding ...

Accelerated Chemistry 4 Chemical Bonding Chapter 6 Pre-test, continued 17.As atoms bond with each other, they a. increase

their potential energy, thus creating less stable arrangements of matter. b. decrease their potential energy, thus creating less stable arrangements of matter.

Chapter 6 PRETEST: Chemical Bonding - Password

Chapter 5 Bonding Test A. Do Not Write on This Test!!! You may use a periodic table. Multiple Choice-Identify the choice that best completes the statement or answers the question. Electrons involved in bonding between atoms are. a. valence electrons b. inside the nucleus. c. closest to the nucleus d. positively charged

Chapter 5 Bonding Test - Dublin Unified School District

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CHAPTER 6 - CHEMICAL BONDING

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6 Chemical Bonding - Effingham County School District Chapter 6 Chemical Bonding Test Review Flashcards | Quizlet

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